

	L(+)-Lactic Acid	ł			REVISION DATE 24/04/07 REF. SD0010/2007-01
1. IDENTIFICATION OF THE SUBSTANCE / PREPARATION	# Product name		PURAC [®] PURAC [®] Sanilac	;	
AND THE	# Use of the Sub	stance	Food additive, S	pecialty che	mical
COMPANY / UNDERTAKING	Supplier		PURAC biochen Arkelsedijk 46 NL-4206 AC Go The Netherlands	orinchem S	PURAC bioquimica Gran Vial 19 - 25 08160 Montmelo-Barcelona Spain
	Telephone		++31 183 69569		++34 93 568 6300
	Fax Emergency tele	phone	++31 183 69560 ++31 183 69569		++34 93 568 3955 ++34 93 568 6300 (Ext 222)
	Supplier		PBR sínteses Rua Augusta, ni São Paulo SP- Brazil	r. 1939 - sal	a 122 / 123
	Telephone		+55 11 3062 15	35	
	Fax		+55 11 3062 40		
	Emergency Tele	pnone	+55 11 3062 15	35	
2. COMPOSITION / INFORMATION ON INGREDIENTS	# Chemical nam the substance	e of	L(+) Lactic Acid	aqueous so	lution
	# Synonyms		Lactic Acid aque	eous solutio	n: S(+)-2-hydroxy propionic acid.
	CAS-No.	79-33-4	EC-No	5. 201-196-2	2
3. HAZARDS IDENTIFICATION	Most important	hazards	Irritating to eyes	and skin. R	tisk of serious damage to eyes.
		May cause irritation of respiratory tract. May cause irritation of the mucous membranes.			•
		Specific hazards			
	Inhalation	(short ar	nd long term)		, breathing difficulties, ne, dizziness
			id long term)		of digestive system
	Skin contact		d long term)	irritation	ritation, blurrad vision
	Eye contact Inge stion		id long term) id long term)		rritation, blurred vision omiting, gastrointestinal disturbance

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For further information:



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4. FIRST AID MEASURES	General advice Inhalation Skin contact Eye contact Ingestion Protection of first-aiders Notes to physician	Show this safety data sheet to the doctor in attendance. Move to fresh air. If symptoms persist, call a physician. Wash off immediately with soap and plenty of water removing all contaminated clothes and shoes. Obtain medical attention. Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Obtain medical attention. Consult a physician. If conscious, drink plenty of water. Never give anything by mouth to an unconscious person. Wear impervious gloves and tightly fitting safety goggles. Avoid contact with skin and eyes. Oxygen, if needed. Avoid gastric lavage.
5. FIRE-FIGHTING MEASURES	Suitable extinguishing me Extinguishing media which not be used for safety reas Specific hazards Special protective equipme for firefighters Specific methods	n must None. sons Thermal decomposition can lead to release of irritating gases and vapors.
6. ACCIDENTAL RELEASE MEASURES	Personal precautions	Avoid contact with skin and eyes. Wear impervious gloves and tightly fitting safety goggles.
	Environmental precautions	Do not flush into surface water or sanitary sewer system.
	Methods for cleaning up	Dam up. Neutralize with limestone powder, lime, soda ash. Soak up with inert absorbent material (e.g. sand, silica gel, acid binder, universal binder, sawdust). Take up mechanically and collect in suitable container for disposal.

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7. HANDLING AND STORAGE	Technical measures/Precautions Safe handling advice	Avoid temperatures above 392°F (200°C). Avoid contact with skin and eyes. Wear impervious gloves and tightly fitting safety goggles. Do not breathe spray mist.
	Technical measures/ Storage conditions Packaging material	Store according to all current regulations. Keep container tightly clos ed. Keep in a dry, cool place. Plastic or stainless steel 316 L containers.
8. EXPOSURE CONTROLS / PERSONAL PROTECTION	Engineering measures to reduce exposure Exposure limit(s)	Insure adequate ventilation, especially in confined areas. None.
	Personal protection equipment Respiratory protection	Not required; except in case of aerosol formation. Breathing apparatus needed only when aerosol or mist is formed.
	Hand protection	Rubber gloves.
	Eye protection Skin and body protection Hygiene measures	Face-shield, tightly fitting safety goggles. Long sleeved clothing, chemical resistant apron boots. Avoid contact with skin When using, do not eat, drink or smoke. Remove and wash contaminated clothing before re-use.
9. PHYSICAL AND CHEMICAL	Form	aqueous solution
PROPERTIES	Color	colorless/yellow/ light brown
	Odor	characteristic
	Molecular Weight pH	90.08 2 @ 77°F (25°C)
	Boiling point/range	230°F (110°C) (40% solution)
	Decomposition temperature	257°F (125°C) (90% solution) > 392 °F (> 200°C)
	Autoignition temperature	none
	Flash point	> 234°F (> 112°C)
	Explosion limits	not applicable
	Density	1190 - 1250 kg/m ³
	Surface tension	50 - 44 mN/m (50 - 90% solution)
	Solubility	Water solubility: completely soluble Partition coefficient (n-octanol/water) log Pow = - 0.62
	Viscosity	5 - 60 mPa.s @ 77°F (25°C) (50 - 90% solution)

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For further information:



10. S R

11. T

SAFETY DATA SHEET

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STABILITY AND REACTIVITY	Stability Conditions to avoid Materials to avoid Hazardous decomposition products	Stable at normal conditions. Avoid temperatures above 392°F (200°C). Oxidizing agents, metals, acids and bases. Carbon oxides. Thermal decomposition can lead to release of irritating gases and vapors.
TOXICOLOGICAL NFORMATION	Acute toxicity	LD50/oral/rat=3730 mg/kg LD50/oral/mouse=4875 mg/kg LD50/dermal/rabbit>2000mg/kg
	Irritation	Eyes-rabbit: severe. Skin guinea pig: slight - none. Skin rabbit: severe. Tests on animals have shown that the effect of lactic acid on skin is species dependent. Human experience and results on guinea pigs have shown that it is irritant
	Local effects	and not corrosive. Irritating to eyes and skin. Risk of serious damage to eyes. Inhalation of mist causes irritation of respiratory system.
	Carcinogen Status Mutagenic Data	None. Tests on bacterial or mammalian cell cultures did not show mutagenic effects.
	Major effects of exposure	
	Inhalation	Inhalation of vapors is irritating to the respiratory system, may cause throat pain and cough. Inhalation of vapors in high concentration may cause shortness of breath (lung oedema). Chronic exposure may cause dermatitis, gastrointestinal disturbance, coughing.
	Skin contact	May cause skin irritation. Prolonged skin contact may produce dermatitis.
	Eye contact	Severe eye irritation. Risk of serious damage to eyes. Liquid causes severe inflammation of conjunctiva and may cause severe damage of the cornea.
	Ingestion	Ingestion may cause gastrointestinal irritation, nausea,
	Further information	vomiting and diarrhea. May cause stomach perforation As an important metabolite in man, animals and plants, it is naturally formed and metabolised.

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12. ECOLOGICAL INFORMATION	Mobility Persistence / degradability	Completely soluble. Readily biodegradable, according to appropriate OECD test. Biochemical oxygen demand $(BOD)_5 = 0.45 \text{ mg } \Omega_2 / \text{mg}$. Biochemical oxygen demand $(BOD)_{20} = 0.60 \text{ mg } \Omega_2 / \text{mg}$. Chemical oxygen demand $(COD) = 0.90 \text{ mg } \Omega_2 / \text{mg}$.
	Bioaccumulation Ecotoxicity	None. EC50/48h/Daphnia = 240mg/l LC50/48h/Fish = 320 mg/l EC50/Algae = 3500 mg/l (neutral)
	Further information	Natural product.
- 13. DISPOSAL CONSIDERATIONS	Waste from residues / unused products	Subject to disposal regulations US EPA 40 CFR 2 62.
	Contaminated packaging	Clean container with water. Empty containers should be taken for local recycling, recovery or waste disposal.
14. TRANSPORT INFORMATION	Not classified as dangerous in the r	neaning of transport regulations.
15. REGULATORY INFORMATION	US Regulations	USA TSCA Inventory Status Y SARA III N California Proposition 65 N Carcinogen status OSHA:N, NTP:N, IARC: N FDA GRAS
	EU Classification	
	Symbols R- Phrases	Xi - Irritant R41 - Risk of serious damage to eyes.
	S-Phrases	 R38 - Irritating to skin. S24 - Avoid contact with skin. S26 - In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. S37/39 - Wear suitable gloves and eye/face protection.
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16. OTHER INFORMATION		1(health)-0(flammability)-0(reactivity) 2(health)-0(flammability)-0(reactivity) B(protective equipment)		
	CAS-No. 50-21-5 (general)	EC-No. 200-018-0 (general)		
	EEC Food additive: E 270			
	Further information on the safety assessment of lactic acid and its salts can be obtained in a CFTA Report of June 6th 1997.			
	Additional data on the calculated ecotoxicity of lactic acid and its salts and esters can be obtained in a report entitled 'The ecotoxicity and biodegradability of lactic acid, alkyl lactate esters and lactic acid salts' by Bowmer et al. (Reference: Chemosphere 37: 1317-1333 (1998))			
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